1. Using your laboratory textbook and your write-up of a similar experiment, **outline a pre-lab**: grams of reactants and solvents, reaction conditions, quenching, work-up, purification, analysis.

2. Bromine comes as a 24% by weight bromine solution in conc HBr (density=1.70g/mL).

3. You will need to use 1.5 equivalents of Br₂.

4. You will need to heat the benzanilide/acetic acid mixture in a 55-60° C water bath until the benzanilide dissolves.

5. You will then turn off the heat and remove the water bath. The Br₂ solution will then be added.